organic papers

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Key indicators

Single-crystal X-ray study T = 298 K Mean σ (C–C) = 0.004 Å R factor = 0.049 wR factor = 0.150 Data-to-parameter ratio = 12.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

3-Benzoylmethyl-2-chloro-N-methylmaleimide

In the title compound, $C_{13}H_{10}CINO_3$, the dihedral angle between the two rings is 68.9 (3)°. The crystal structure is stabilized by weak intermolecular $C-H \cdots O$ hydrogen bonds.

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Comment

We have investigated the photo-induced reaction of 3,4dichloromaleimide (3,4-dichloro-pyrrole-2,5-dione) and 2trimethylsiloxyphenylethene, and obtained the title compound, (I), as one of the products. An X-ray crystallographic analysis was undertaken to establish its structure and configuration.



In the structure of (I) (Fig. 1), bond lengths and angles are in good agreement with the expected values (Allen *et al.*, 1987). Both the maleimide ring and the phenyl ring are planar to within ± 0.02 Å. The carbonyl group linked to the phenyl ring is nearly in the plane of the phenyl ring, with atom C7 deviating by 0.019 (3) Å. Similarly, atom Cll is in the plane of the maleimide ring, with a deviation of 0.045 (3) Å. The deviations of atom C8 from the phenyl and maleimide rings are 0.058 (2) and 0.100 (2) Å, respectively. Finally, the dihedral angle between the phenyl and maleimide rings is 68.9 (3)°.

In the packing of the title compound, the molecules are linked into layers (Fig. 2) by $C-H\cdots O$ intermolecular contacts (Table 2). No significant $\pi-\pi$ interactions are observed in the crystal structure.

Experimental

The title compound was one of the products of a photo-induced reaction between 3,4-dichloromaleimide (0.5 mol l⁻¹) (3,4-dichloropyrrole-2,5-dione) and an excess amount of 2-trimethyl-siloxyphenylethene (1.5 mol l⁻¹) in benzene. The solution was irradiated for 10 h with UV light of $\lambda > 300$ nm.The title compound was isolated using column chromatography (yield 35%). Single crystals suitable for X-ray diffraction analysis were obtained by slow evaporation of a petroleum ether–ethyl acetate (4:1) solution.

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Crystal data

 $C_{13}H_{10}CINO_3$ $M_r = 263.67$ Monoclinic, P_{2_1}/n a = 5.474 (2) Å b = 24.185 (3) Å c = 9.283 (2) Å $\beta = 98.78$ (3)° V = 1214.4 (5) Å³ Z = 4

Data collection

Enraf–Nonius CAD-4 diffractometer $\omega/2\theta$ scans Absorption correction: ψ scan (XCAD4; Harms & Wocadlo, 1995) $T_{min} = 0.885$, $T_{max} = 0.965$ 2298 measured reflections 2085 independent reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.049$ $wR(F^2) = 0.150$ S = 1.032085 reflections 163 parameters H-atom parameters constrained

Table 1

Selected geometric parameters (Å, °).

O1-C7	1.211 (3)	C7-C8	1.516 (4)
C5-C7	1.481 (4)	C12-Cl	1.700 (3)
C5-C7-C8 C9-C8-C7	118.5 (3) 112.6 (3)	C11-C12-Cl	120.4 (2)
C6-C5-C7-C8 C5-C7-C8-C9	3.5 (4) 169.7 (3)	C7-C8-C9-C10 N-C11-C12-Cl	-68.3(4) -179.2(2)

 $D_x = 1.442 \text{ Mg m}^{-3}$

Cell parameters from 25

 $0.40 \times 0.40 \times 0.10 \ \mathrm{mm}$

1321 reflections with $I > 2\sigma(I)$

Mo $K\alpha$ radiation

reflections

 $\theta=1.7{-}25.0^\circ$

 $\mu = 0.31 \text{ mm}^{-1}$

T = 298 (2) K

 $R_{\rm int} = 0.066$

 $\theta_{\rm max} = 25.0^{\circ}$

 $h = 0 \rightarrow 6$

 $k = 0 \rightarrow 28$

 $l = -11 \rightarrow 10$

 $(\Delta/\sigma)_{\text{max}} = 0.003$ $\Delta\rho_{\text{max}} = 0.32 \text{ e} \text{ Å}^{-3}$

 $\Delta \rho_{\rm min} = -0.26 \text{ e} \text{ } \text{\AA}^{-3}$

(Sheldrick, 1997)

3 standard reflections

every 200 reflections

intensity decay: none

 $w = 1/[\sigma^2(F_o^2) + (0.0337P)^2 + P]$

Extinction correction: SHELXTL

Extinction coefficient: 0.022 (3)

where $P = (F_o^2 + 2F_c^2)/3$

Plate, colourless

Table 2

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdot \cdot \cdot A$
$\begin{array}{c} C3 - H3A \cdots O3^{i} \\ C8 - H8B \cdots O1^{ii} \end{array}$	0.93 0.97	2.59 2.51	3.499 (5) 3.375 (5)	163 147
Symmetry codes: (i) -	$-x - \frac{1}{2}, y + \frac{1}{2}, -z$	$x + \frac{1}{2}$; (ii) $x + \frac{1}{2}$,	$-y + \frac{1}{2}, z + \frac{1}{2}.$	

Symmetry codes. (i) $-x - \frac{1}{2}$, $y + \frac{1}{2}$, $-z + \frac{1}{2}$, (ii) $x + \frac{1}{2}$, $-y + \frac{1}{2}$, $z + \frac{1}{2}$.

H atoms were positioned geometrically and treated as riding on their parent C atoms, with C–H distances constrained to 0.93 (aromatic CH), 0.97 (methylene CH₂) or 0.96 Å (methyl CH₃), and with $U_{\rm iso}(\rm H) = 1.2U_{eq}(\rm C)$ for aromatic CH and methylene CH₂, or $1.5U_{eq}(\rm C)$ for the methyl group.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*;



Figure 1

The structure of (I), showing 30% probability displacement ellipsoids and the atom-numbering scheme.



Figure 2

The crystal structure of (I), viewed along [100].

molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2003).

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References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1–19.

Enraf-Nonius (1989). CAD-4 Software. Version 5.0. Enraf-Nonius, Delft, The Netherlands.

Harms, K. & Wocadlo, S. (1995). XCAD4. University of Marburg, Germany. Sheldrick, G. M. (1997). SHELXTL. Version 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.

Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.